COURSE OUTLINE

(1) GENERAL

SCHOOL	School of Environment			
ACADEMIC UNIT	Department of Environment			
LEVEL OF STUDIES	Undergraduate			
COURSE CODE	239Y SEMESTER 3			
COURSE TITLE	Organic Chemistry			
INDEPENDENT TEACHING ACTIVITIES if credits are awarded for separate components of the course, e.g. lectures, laboratory exercises, etc. If the credits are awarded for the whole of the course, give the weekly teaching hours and the total credits		WEEKLY TEACHING HOURS	CREDITS	
	Lectures 3			
	Laboratory exercises			
TOTAL			3	4
Add rows if necessary. The organisation of teaching and the teaching methods used are described in detail at (d).				
COURSE TYPE general background, special background, specialised general knowledge, skills development	General back	ground		
PREREQUISITE COURSES:	Chemistry			
LANGUAGE OF INSTRUCTION and EXAMINATIONS:	Greek			
IS THE COURSE OFFERED TO ERASMUS STUDENTS	No			
COURSE WEBSITE (URL)	https://www.env.aegean.gr/all_courses/organic-chemistry/			

(2) LEARNING OUTCOMES

Learning outcomes

The course learning outcomes, specific knowledge, skills and competences of an appropriate level, which the students will acquire with the successful completion of the course are described.

Consult Appendix A

- Description of the level of learning outcomes for each qualifications cycle, according to the Qualifications Framework of the European Higher Education Area
- Descriptors for Levels 6, 7 & 8 of the European Qualifications Framework for Lifelong Learning and Appendix B
- Guidelines for writing Learning Outcomes
 - Provide a basic understanding of organic chemicals.
 - Write condensed structures of organic molecules.
 - Determine which structures represent different or similar molecules.
 - Determine and name and structure of organic compounds.
 - Classify molecules as alkanes, alkenes, alkynes, alcohols, ethers, ketones, esters, carboxylic acids, amines, aldehydes or aromatic compounds and state their properties.

General Competences

Taking into consideration the general competences that the degree-holder must acquire (as these appear in the Diploma				
supplement and appear below), at which of the following does the course aim?				
Search for, analysis and synthesis of data and information,	Project planning and management			
with the use of the necessary technology	Respect for difference and multiculturalism			
Adapting to new situations	Respect for the natural environment			
Decision-making	Showing social, professional and ethical responsibility and			
Working independently	sensitivity to gender issues			
Teamwork	Criticism and self-criticism			

Working in an international environment Working in an interdisciplinary environment Production of new research ideas Production of free, creative and inductive thinking

Search for, analysis and synthesis of data and information, with the use of the necessary technology Adapting to new situations Decision-making Production of new research ideas

Others...

(3) SYLLABUS

- 1. Introduction to Organic Chemistry
- 2. Chemical structures and bonds
 - 3. Organic reactions-nomenclature
 - 4. Saturated hydrocarbons
- 5. Unsaturated hydrocarbons
- 6. Aromatic hydrocarbons
- 7. Haloalkanes, alcohols, ethers
- 8. Carbonyl compounds
- 9. Stereochemistry
- 10. Structure determination
- 11. Biomolecules
- 12. Seminar

(4)

(5)

(6) TEACHING and LEARNING METHODS - EVALUATION

DELIVERY Face-to-face, Distance learning, etc.	Face-to-face		
USE OF INFORMATION AND COMMUNICATIONS TECHNOLOGY Use of ICT in teaching, laboratory education, communication with students	Presentations, Power Point, Microsoft Office, Internet		
TEACHING METHODS	Activity	Semester workload	
The manner and methods of teaching are	Lectures	39	
Lectures, seminars, laboratory practice,	Study and analysis of	78	
fieldwork, study and analysis of bibliography,	bibliography		
tutorials, placements, clinical practice, art workshop interactive teaching educational			
visits, project, essay writing, artistic creativity,			
etc.			
The student's study hours for each learning			
activity are given as well as the hours of non-			
ECTS			
	Total	117	
STUDENT PERFORMANCE	Language of evaluation: Greek		
EVALUATION			
Description of the evaluation procedure	Elective intermediate exam (40%) and written final		
Language of evaluation, methods of evaluation,	exam (60%), OR		
questionnaires, short-answer questions, open-	Written final exam (100%).		
ended questions, problem solving, written work,			
essay/report, oral examination, public presentation. laboratory work clinical	Evaluation methods:		
examination of patient, art interpretation, other	Short Answer Questions 25%		
Specifically defined avaluation criteria are given	Open Questions 75%		
and if and where they are accessible to students.			

(7) ATTACHED BIBLIOGRAPHY

- Suggested bibliography:

McMurry (2017). Οργανική Χημεία. 1η έκδοση. Πανεπιστημιακές Εκδόσεις Κρήτης

- Related academic journals: